Refinement	
Refinement on F	$\Delta \rho_{\rm min} = -1.52 \ {\rm e} \ {\rm \AA}^{-3}$
Final $R = 0.0298$	Extinction correction: $F^* =$
wR = 0.0213	$F(1+0.002xF^2/\sin 2\theta)^{-1/4}$
S = 1.21	Extinction coefficient: $x =$
316 reflections	0.0028 (6)
16 parameters	Atomic scattering factors
$w = 1/[\sigma^2(F) + 0.0001F^2]$	from International Tables
$(\Delta/\sigma)_{\rm max} \leq 0.001$	for X-ray Crystallography
$\Delta \rho_{\rm max} = 1.33 \ {\rm e} \ {\rm \AA}^{-3}$	(1974, Vol. IV)

Table 1. Fractional atomic coordinates and equivalent isotropic displacement coefficients (Å²)

 U_{eq} is defined as one third of the trace of the orthogonalized U_{ij} tensor. The anisotropic displacement exponent takes the form $-2\pi^2(h^2a^{\star 2}U_{11})$ $+ ... + 2hka^{\star}b^{\star}U_{12})$.

x	у	z	U_{eq}
	•		-
0.4874 (1)	0.1216(1)	0.2500	0.06(1)
0.3474 (1)	0.2971 (1)	0.2500	0.06(1)
0.3333	0.6666	0.2500	0.05 (1)
0.4874 (1)	0.1215 (1)	0.2500	0.04 (1)
0.3472 (1)	0.2969 (1)	0.2500	0.04 (1)
0.3333	0.6666	0.2500	0.04 (1)
0.4875 (1)	0.1214(1)	0.2500	0.03 (1)
0.3473 (1)	0.2967 (1)	0.2500	0.03 (1)
0.3333	0.6666	0.2500	0.03 (1)
	x 0.4874 (1) 0.3474 (1) 0.3333 0.4874 (1) 0.3472 (1) 0.3333 0.4875 (1) 0.3473 (1) 0.3333	x y 0.4874 (1) 0.1216 (1) 0.3474 (1) 0.2971 (1) 0.3333 0.6666 0.4874 (1) 0.1215 (1) 0.3472 (1) 0.2969 (1) 0.3333 0.6666 0.4875 (1) 0.1214 (1) 0.3473 (1) 0.2967 (1) 0.3333 0.66666	xyZ 0.4874 (1) 0.1216 (1) 0.2500 0.3474 (1) 0.2971 (1) 0.2500 0.3333 0.6666 0.2500 0.4874 (1) 0.1215 (1) 0.2500 0.3472 (1) 0.2969 (1) 0.2500 0.3333 0.6666 0.2500 0.4875 (1) 0.1214 (1) 0.2500 0.3473 (1) 0.2967 (1) 0.2500 0.3333 0.6666 0.2500

Table 2. Selected bond lengths (Å) for V₃S₄ determined at different temperatures

	55	•	
	295 K	170 K	100 K
V2V	2.864(1)	2.859(1)	2.856(1)
V2V	3.120(1)	3.115(1)	3.114(1)
V-2S(2)	2.444(1)	2.440(1)	2.440(1)
V - S(1)	2.358(1)	2.355(1)	2.354(1)
V - S(1)	2.497(1)	2.493(1)	2.490(1)
V - 2S(1)	2.340(1)	2.337(1)	2.338(1)
VS(av.)	2.404	2.400	2.400

Lists of structure factors, anisotropic displacement parameters and selected bond angles have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 55842 (9 pp.). Copies may be obtained through The Technical Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England. [CIF reference: SH1019]

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A New Type of One-Dimensional Compound: Structure of Nb₄(Te₂)₄Te₄I

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Abstract

The new infinite-chain niobium telluride iodide has been prepared by reaction of the elements at 893 K. Nb₄(Te₂)₄Te₄I represents a new one-dimensional structure type. The structure consists of $[Nb_4(Te_2)_4-Te_4I]_{\infty}$ chains which are formed by the four-nuclear butterfly cluster units 'Nb₄(Te₂)₄Te₄' with the I atoms bridging between different cluster units.

Comment

Low-dimensional niobium and tantalum chalcogenide compounds have received much attention in the last few years (Rouxel, 1986a; Badding & Disalvo, 1990; Liimatta & Ibers, 1987, 1988, 1989). The main interests lie in the synthesis, structure and especially the unusual properties such as superconductivity (Shenoy, Dunlap & Fradin, 1981), chargedensity waves (Rouxel, 1986b; Mihaly, Chen, Kim & Gruner, 1988), anisotropic electrical conductivity and optical behaviour (Wilson, Disalvo & Mahajan, 1975). Although numerous investigations have been conducted on niobium/tantalum chalcogenides, examples of low-dimensional niobium/tantalum telluride compounds are still very limited. Here we report on the structure determination of a new onedimensional compound Nb4(Te2)4Te4I. The compound was obtained by solid-state reactions of the approximate stoichiometric elements at about 893 K. Single crystals used in the structure determination

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were obtained by vapour transport methods with iodine as the transport agent. Details of synthetic work and physical properties, along with a theoretical consideration will be published elsewhere (Shuiquan, Honghui, Jinshun & Jingling, 1992).

A perspective view of the structure along the c axis is shown in Fig. 1. The atom-labelling scheme is illustrated in Fig. 2. As shown in Fig. 1, the structure of the title compound is based on the Nb₄(Te₂)₄Te₄I monomers, stacked in a head-to-tail manner to form the [Nb₄(Te₂)₄Te₄I]_{∞} chains, which are held together by van der Waals interactions and pack in a monoclinic unit cell. One I atom bridges two different 'Nb₄(Te₂)₄Te₄' cluster units. Nb atoms in the cluster unit bond together to form the four-nuclear butterfly cluster core, to our knowledge hitherto not found in the niobium or tantalum cluster chemistry.

Te atoms coordinate the Nb atoms in two ways: one in the form of Te_2 dimers which coordinate over the edge of the Nb₄ butterfly core; the other simply as the terminal coordinating atoms; this leads to the formula Nb₄(Te₂)₄Te₄I.



Fig. 1. Perspective view of the structure of Nb₄(Te₂)₄Te₄I along the c axis. Atoms corresponding to z < 0.5 have been omitted.



Fig. 2. Projection of a 'Nb₄(Te₂)₄Te₄I' monomer along the b axis, showing the atom labelling.

The shortest Nb-Nb distance in the cluster core averages 3.046 (3) Å, and is consistent with the reported values of Nb-Nb bond lengths [2.986 Å, Nb₄Se₄I₄ (Ben Yaich, Jegaden, Potel & Sergent, 1984)]. The long distances between the Nb atoms at the opposite corners of the butterfly core (3.765 Å average) indicate no chemical bonding and suggest that the core is a butterfly rather than a tetrahedron. The Nb—Te bond lengths range from 2.771 (3) to 2.889 (2) Å in the 'Nb₄(Te₂)₄Te₄' cluster unit. These distances are similar to the values found in NbTe₄ [2.902 Å (Selte & Kjekshus, 1964)]. The Te-Te bond lengths range from 2.705 (3) to 2.717 (3) Å, which are shorter than found in NbTe₄ [2.867 Å (Selte & Kjekshus, 1964)], and in Te crystals [2.835 (2) Å (Cherin & Unger, 1967)]. The Nb-I distance is 2.998 Å (average), which is a little larger than the Nb— I^{a-a} distance [2.958 (3) Å] in Nb₆I₁₁ (Imoto & Simon, 1982), and indicates weaker Nb-I bonding. The weak Nb-I bonding shows that the 'Nb₄(Te₂)₄Te₄' cluster units are relatively independent with respect to each other. Therefore, the title compound may be viewed as in the molecular state with the formula $Nb_4(Te_2)_4Te_4I$. The title compound is indeed much different from other families of onedimensional compounds that have extended metalmetal interactions, for example: Ta_4ZTe_4 (Z = stabilizing heteroatom) (Badding & Disalvo, 1990); $(MX_4)_n Y$ (M = Nb, Ta; X = S, Se; Y = Br, I)(Gressier, Meerschaut, Guemas, Rouxel & Monceau, 1984). A new member belonging to the same family, $Ta_4(Te_2)_4Te_4I$, has been synthesized. The structure determination is in progress.

In summary, the title compound represents a new one-dimensional class with typical molecular characteristics, especially the novel Nb_4/Ta_4 butterfly cluster core. Further studies of the properties of these phases, through breaking the I-atom bridges and/or substituting I atoms for other ligands, are being conducted.

Experimental

Crystal data

 $Nb_4(Te_2)_4 Te_4I$ $M_r = 2029.73$ Monoclinic C2/c

a = 21.957 (5) Å

b = 6.147 (3) Å c = 19.925 (4) Å

 $\beta = 122.40 (1)^{\circ}$

 $V = 2270.6 \text{ Å}^3$

Z = 4

 $D_m = 5.8$ (5) Mg m⁻³ Density measured by micropycnometry in toluene Mo radiation $\lambda = 0.71073 \text{ Å}$ Cell parameters from 25 reflections $\theta = 14-15^{\circ}$ $\mu = 18.38 \text{ mm}^{-1}$ T = 296 KRectangular $0.12 \times 0.08 \times 0.06 \text{ mm}$ Black Crystal source: vapour transport methods

Data collection Enraf-Nonius CAD-4 1375 observed reflections diffractometer $[I > 3\sigma(I)]$ $\omega/2\theta$ scans of rate 1-5° $R_{\rm int} = 0.038$ min^{-1} and width (0.45 $\theta_{\rm max} = 25.0^{\circ}$ + $0.35 \tan\theta$)° $h = 0 \rightarrow 26$ Absorption correction: $k = 0 \rightarrow 7$ empirical (North, Phillips $l = -23 \rightarrow 19$ 3 standard reflections & Mathews, 1968) $T_{\rm min} = 0.53, T_{\rm max} = 1.00$ frequency: 60 min 2269 measured reflections intensity variation: 2.2% 2202 independent reflections

Refinement

Refinement on F	$\Delta \rho_{\rm max}$ = 0.86 e Å ⁻³
Final $R = 0.052$	$\Delta \rho_{\rm min}$ = 0.81 e Å ⁻³
wR = 0.062	Extinction correction:
S = 1.26	Zachariasen (1963)
1375 reflections	Extinction coefficient: 0.61×10^{-6}
79 parameters	Atomic scattering factors
$w = 1/[\sigma^2(F_o) + (0.03 F_o)^2 + 1.0]$	from International Tables for X-ray Crystallography
$(\Delta/\sigma)_{\rm max}$ = 0.01	(1974, Vol. IV)

 Table 1. Fractional atomic coordinates and equivalent isotropic thermal parameters (Å²)

 $B_{\rm eq} = \frac{4}{3} \sum_i \sum_j \beta_{ij} \mathbf{a}_i . \mathbf{a}_j.$

	x	у	z	Bea
1	0.0000	-0.3225 (4)	0.7500	4.35 (6)
Nb(1)	0.07816(6)	0.2970 (2)	0.73611 (6)	0.53 (2)
Nb(2)	-0.06469 (6)	0.0562 (2)	0.63875 (6)	0.59 (3)
Te(1)	-0.1432 (1)	-0.1965 (4)	0.5011(1)	3.32 (5)
Te(2)	-0.02309 (9)	0.3595 (4)	0.56782 (9)	3.00 (4)
Te(3)	0.18146 (9)	0.0047 (4)	0.8362 (1)	3.01 (4)
Te(4)	0.17261 (9)	0.3635 (4)	0.9049 (1)	2.94 (4)
Te(5)	0.17463 (9)	0.5548 (4)	0.7199(1)	3.36 (5)
Te(6)	0.05870 (9)	-0.0065 (4)	0.62339 (9)	3.03 (4)

Table 2. Geometric parameters (Å, °)

Nb(1)Nb(2)	3.045 (2)	$Nb(1)-Nb(2)^{i}$	3.047 (3)	
Nb(1)—Te(2)	2.889 (2)	Nb(1)—Te(3)	2.777 (2)	
Nb(1)—Te(4)	2.882 (2)	Nb(1)—Te(5)	2.800 (3)	
Nb(1)—Te(6)	2.771 (3)	Nb(2)Te(1)	2.800 (2)	
Nb(2)—Te(2)	2.774 (3)	Nb(2)—Te(6)	2.911 (3)	
$Nb(2)$ — $Te(3)^{i}$	2.887 (3)	Nb(2)—Te(4) ⁱ	2.780 (2)	
Te(3)—Te(4)	2.705 (3)	Te(2)—Te(6)	2.717 (3)	
1—Nb(2)	2.996 (2)	I—Nb(1) ⁱⁱ	3.000 (3)	
Te(2)—Nb(1)—Te(3)	138.31 (8)	Te(2)—Nb(1)—Te(5)	87.13 (7)	
Te(3)Nb(1)Te(4)	57.09 (6)	Te(4)-Nb(1)-Te(5)	86.49 (6)	
Te(2)—Nb(1)—Te(6)	57.33 (7)	Te(4)—Nb(1)—Te(6)	138.02 (7)	
Te(3)—Nb(1)—Te(5)	94.79 (8)	Te(3)-Nb(1)-Te(6)	81.05 (7)	
Te(2)—Nb(1)—Te(4)	163.91 (8)	Te(5)-Nb(1)-Te(6)	94.87 (8)	
Te(1)—Nb(2)—Te(2)	94.12 (7)	$Te(1) - Nb(2) - Te(3)^{i}$	87.69 (8)	
Te(1)—Nb(2)—I	95.32 (7)	$Te(1) - Nb(2) - Te(4)^{i}$	95.52 (6)	
Te(1)—Nb(2)—Te(6)	86.73 (7)	$Te(2) - Nb(2) - Te(3)^{i}$	137.81 (7)	
Te(2)—Nb(2)—I	138.10 (6)	$Te(2) - Nb(2) - Te(4)^{i}$	80.93 (7)	
Te(2)—Nb(2)—Te(6)	57.04 (7)	$Te(3)^i - Nb(2) - I$	83.34 (5)	
$Te(3)^{i}$ —Nb(2)—Te(4) ⁱ	56.99 (7)	$Te(3)^{i}$ -Nb(2)-Te(6)	164.58 (8)	
$I - Nb(2) - Te(4)^{i}$	138.24 (8)	I = Nb(2) = Te(6)	82.87 (6)	
$Te(4)^{i}$ —Nb(2)—Te(6)	137.95 (8)	$Nb(2) - I - Nb(2)^{i}$	78.03 (7)	
$Nb(1)^n - I - Nb(1)^m$	77.55 (8)	$Nb(2)-Nb(1)-Nb(2)^{i}$	76.51 (5)	
$Nb(1)-Nb(2)-Nb(1)^{i}$	76.18 (5)			
Symmetry code: (i) $-x, y, \frac{3}{2} - z$; (ii) $x, y - 1, z$; (iii) $-x, y - 1, \frac{3}{2} - z$.				

The crystals of $Nb_4(Te_2)_4 Te_4I$ are black and show metallic lustre. Microprobe analysis of the single crystals was made with

the Jeol scanning electron microscope. The experimental values are: Nb 19.6(7), Te 75.2(4), I 5.1(7) wt %. Calculated for Nb₄(Te₂)₄Te₄I: Nb 18.31, Te 75.44, I 6.25 wt %. These results are within the accuracy of the microprobe of $\pm 20\%$. Data collection and cell refinement: Enraf-Nonius (1977) CAD-4 diffractometer control software. Programs used to solve structure: *MULTAN*11/82 (Main, Fiske, Hull, Lessinger, Germain, Declercq & Woolfson, 1982). Programs used for full-matrix least-squares refinement and other calculations: Enraf-Nonius *SDP* package (Frenz, 1978). Calculations were performed on a VAX 785 computer.

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Lists of structure factors, anisotropic thermal parameters and complete geometry have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 55906 (14 pp.). Copies may be obtained through The Technical Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England. [CIF reference: AB1041]

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